YEAR 10 (A-E)— CHEMISTRY

WEEK 6 (26th April to 30th April)

Work sent to the students through Google classroom / Zoom Learning Platform/ Whatsapp group

Lessson Objective: Revision of topics flame tests and photometry, Tests for positive ions and negative ions.

Resources: Text book, Worksheet file, video, power point presentations.

Sunday – zero period(boys) Sunday – 8 th period (girls)	Reinforcement of flame tests and photometry. Powerpoint presentation contains interactive questions.
Monday - 5 th and 6 th period (Girls) Tuesday - 3 rd and 4 th period (Boys)	Revise precipitation reactions for the identification of positive and negative ions.
	Write balanced equations and ionic equations for all the reactions involved.
	Solve worksheet file questions.
Tuesday - 8 th period (Girls)	
Monday – 8 th period (Boys)	Practice descriptive type questions from past papers.

YEAR 10 G/H – CHEMISTRY (IGCSE)

WEEK 6 (26th April to 30th April)

Work Sent to the students through Group email/ Google classroom

Topic 2 – Gases in the atmosphere

Lessson Objective: The combustion of elements in oxygen.

Resources: Text book, Worksheet, GCSE science free lesson video, power point.

Sunday – 7 th and 8 th (Girls & Boys)	To understand how to determine the percentage by volume of oxygen in air using experiments involving reaction of iron and phosphorous with air. *Powerpoint presentation contains interactive questions.* Read pg. 139-140 of text bk. Answer Q2 (a-b) on pg. 144 of the text bk Write answers to page 43 of worksheet file.
Monday -1st (Girls & Boys)	Describe the combustion of elements in oxygen, including magnesium, hydrogen & sulphur. Describe the formation of carbon dioxide from the thermal decomposition of metal carbonates, including copper(II) carbonate. PowerPoint presentation contains interactive questions. Read pg. 141-142 of text bk. Answer Q3(a-h) on pg. 144 of the text bk Write answers to page 44 of worksheet file.
Tuesday - 3 rd period (Girls & Boys)	Know that carbon dioxide is a greenhouse gas and that increasing amounts may contribute to climate change. *PowerPoint presentation contains interactive questions.** Read pg. 142-143 of text bk. Answer Q4 on pg. 144 of the text bk Write answers to page 45 of worksheet file.