

YEAR 10 A/D/E–CHEMISTRY (Girls)

WEEK 3 - (13th Sept to 17th Sept)

Work Sent to the students through Google classroom/ Zoom Learning Platform

Topic: Conservation of mass

Resources: Text book, Worksheet, power point.

| Date | Lesson | Topic | Mode of Teaching | |
|------------------------|--------|---|--------------------------|---|
| 13/9/2020 Sunday | 3 | Learning Objective: Calculate the concentration of solutions in g dm^{-3} Learning outcome: <ul style="list-style-type: none">• Define concentration of solutions.• Use the formula $C = m/V$ to calculate the concentration in g/dm^3 | Zoom | Teacher uses powerpoint presentation that contains the method to calculate the concentration of solution. |
| 16/9/2020 Wednesday | 3 | Learning Objective: Explain the law of conservation of mass applied to: (a) a closed system including a precipitation reaction in a closed flask (b) a non-enclosed system including a reaction in an open flask that takes in or gives out a gas. Learning Outcome: <ul style="list-style-type: none">• State Law of conservation of mass.• Define precipitation reaction.• Compare the difference between a closed and non-enclosed system. | Zoom | Teacher uses powerpoint presentation that contains interactive questions. |
| 17/9/2020 Thursday | 2 3 | Learning Objective: Calculate the masses of reactants and products from balanced equations, given the mass of one substance using mole method . Learning Outcome: Calculate the mass of product formed from a given mass of reactant, using a balanced equation. | Zoom GC | Teacher uses powerpoint presentation that contains the steps to calculate the masses of reactants and products. Instruction will be given in the Google classroom to complete the worksheet questions. |

Homework : Complete the textbook questions SC9b page74

YEAR 10 B/C/F–CHEMISTRY (Boys)

WEEK 3 (13th Sept to 17th Sept)

Work Sent to the students through Google classroom/ Zoom Learning Platform

Topic: Conservation of mass

Resources: Text book, Worksheet, power point.

| Date | Lesson | Topic | Mode of Teaching | |
|------------------------|--------|---|------------------|---|
| 13/9/2020 Sunday | 0 | Learning Objective: Calculate the concentration of solutions in g dm^{-3} Learning outcome: <ul style="list-style-type: none">• Define concentration of solutions.• Use the formula $C = m/V$ to calculate the concentration in g/dm^3 | Zoom | Teacher uses powerpoint presentation that contains the method to calculate the concentration of solution. |
| 14/9/2020 Monday | 1&2 | Learning Objective: Explain the law of conservation of mass applied to: (c) a closed system including a precipitation reaction in a closed flask (d) a non-enclosed system including a reaction in an open flask that takes in or gives out a gas. Learning Outcome: <ul style="list-style-type: none">• State Law of conservation of mass.• Define precipitation reaction. Compare the difference between a closed and non-enclosed system. | Zoom | Teacher uses powerpoint presentation that contains the steps to calculate the masses of reactants and products. |
| 16/9/2020 Wednesday | 4 | Learning Objective: Calculate the masses of reactants and products from balanced equations, given the mass of one substance using mole method . Learning Outcome: Calculate the mass of product formed from a given mass of reactant, using a balanced equation. | GC | Instruction will be given in the Google class room to complete the worksheet questions. |

Homework : Complete the textbook questions SC9b page74