## YEAR 9 A- F - CHEMISTRY

## WEEK 7 (11<sup>th</sup> Oct to 15<sup>th</sup> Oct)

## Work Sent to the students through Group email/ Google classroom **Topic:**– **Chromatography (continuation) and Distillation**

**Resources:** Text book, Worksheet, Board works, GCSE science free lesson video, power point.

Date	Lesson	Торіс	Mode of Teaching	
11 <sup>th</sup> Oct Sunday ( <b>girls</b> )	6	Learning Objective : Interpret a paper chromatogram • to distinguish between pure and impure substances	Zoom	
		<ul> <li>to identify substances by comparison with known substances</li> <li>to identify substances by calculation and use of Rf values</li> </ul>		PPT and Video on Chromatogra phy
12 <sup>th</sup> Oct Monday ( <b>boys</b> )	7	<ul> <li>Success Criteria:</li> <li>Identify pure substances and mixtures on chromatograms.</li> <li>Identify substances that are identical on chromatograms.</li> <li>Calculate Rf values of different components in the chromatogram.</li> <li>Identify the areas where we use chromatographic technique.</li> <li>Suggest some uses of chromatography.</li> </ul>		
12 <sup>th</sup> Oct Monday ( <b>girls</b> )	5	Learning Objective : Reinforcement of concepts on Chromatography Success Criteria:	GC	Worksheet on Chromatogra phy
12 <sup>th</sup> Oct Monday– ( <b>boys</b> )	8	<ul> <li>Describe Chromatography as a separation technique to separate the dissolved salts in a solution.</li> <li>To differentiate between pure and impure substances</li> <li>To calculate the R<sub>f</sub> value</li> </ul>		
12 <sup>th</sup> Oct Monday ( <b>girls</b> )	6	<ul> <li>Learning Objective :</li> <li>Explain the experimental techniques for separation of mixtures by</li> <li>simple distillation</li> <li>fractional distillation</li> </ul>	Zoom	PPT / Video on Simple and Fractional Distillation

14 <sup>th</sup> Oct	1	Success Criteria:	
Wednesday – ( <b>boys</b> )		<ul> <li>Describe how to carry out, and explain what happens in, simple distillation.</li> <li>Describe how to carry out fractional distillation.</li> <li>Compare the similarities and differences between simple and the fractional distillation.</li> <li>Identify when fractional distillation should be used to separate a mixture</li> <li>Suggest the safety precautions needed to reduce risk in distillation experiment.</li> </ul>	

**<u>Homework :</u>** Complete the S1 and E1 questions on page # 9 of the Textbook.