## YEAR 9 A- F – CHEMISTRY

## WEEK 13 (22<sup>nd</sup> Nov to 26<sup>th</sup> November)

Work Sent to the students through Group email/ Google classroom

**Topic:**— SC4c – Electronic Configuration and the Periodic Table

Resources: Text book, Worksheet, Board works, GCSE science free lesson video, power point.

Date	Lesson	Торіс	Mode of Teaching	
22 <sup>nd</sup> Nov Sunday (girls)	6	Learning Objective:  • Predict the electronic configurations of the first 20 elements in the periodic table as diagrams and in the form, for example, 2.8.1  Success Criteria:		PPT/Video on
23 <sup>rd</sup> Nov Monday ( <b>boys</b> )	7	<ul> <li>Draw electron arrangements for elements up to number 20.</li> <li>Predict electron arrangements for first 10 elements</li> <li>Draw shell diagram of first 20 elements.</li> </ul>	Zoom	Electronic Configuration
23 <sup>rd</sup> Nov Monday ( <b>girls</b> )	5	<ul> <li>Learning Objective:</li> <li>Explain how the electronic configuration of an element is related to its position in the periodic table</li> </ul>	Zoom	PPT and Video on Electronic Configuration
23 <sup>rd</sup> Nov Monday– ( <b>boys</b> )	8	<ul> <li>Success Criteria:</li> <li>Identify the group number from the electronic configuration.</li> <li>Extract information from charts and graphs including patterns in the properties of elements</li> <li>Correlate the similarity in the properties with the group number.</li> <li>Use the periodic table to write the electronic configuration.</li> <li>Explain the links between an element's position in the periodic table and its electronic configuration.</li> </ul>		
23 <sup>rd</sup> Nov Monday (girls) 25 <sup>th</sup> Nov Wednesday – (boys)	6	<ul> <li>Learning Objective:</li> <li>Identify elements as metals or non-metals according to their position in the periodic table</li> <li>Success Criteria:</li> <li>Use patterns in the Periodic Table to interpret data and predict few properties of elements.</li> <li>Divide the elements into metals, non metals and semimetals.</li> </ul>	GC	Worksheet SC4c

• Locate the position of metals and non metals in the in the periodic table.	